Calendar portion revised 8-22-13

General Chemistry 102 - Fall 2013

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Lecture	Tues & Thurs	8:30 – 9:45 a.m.	Cudahy Hall-202 (001)
Discussion	Fri	12:35-1:25 p.m.	Flanner Hall-105 (Sect 002)
	Fri	1:40-2:30 p.m.	Flanner Hall-105 (Sect 003)

Office Hours Wed 10:30 a.m. – 12:00 p.m. & Fri 2:00 p.m. – 3:30 p.m.

Required Text: Brown, LeMay, Bursten, Murphy, Woodward Chemistry-The Central Science 12th Ed.
ISBN 978-0-321-69672-4

1. *Exam Dates (subject to change):*

Thursday, September 19, 2013:	**Mid-term Exam 1**
Thursday, October 24, 2013:	**Mid-term Exam 2**
Thursday, November 21, 2013:	**Mid-term Exam 3**
Saturday, December 14, 2013:	**Final Exam, 9:00-11:00 a.m.**

2. *Exams and Grading:*

There are three 75-minute mid-term exams and one 2-hour final exam. The lowest of the three mid-term exams will be dropped. If you miss an hourly exam, that is the exam that will be dropped. No make-up mid-term exams will be given under any circumstances. The final exam is cumulative and cannot be dropped.

MasteringChemistry Homework	50 points	
Discussion	48 points	
Mid-term exam	100 points	(Best two out of three mid-term exams)
Mid-term exam	100 points	
Final Exam	150 points	
TOTAL	448 points	

You must bring a form of photo identification, such as your Loyola Student ID or your driver's license, with you to the exam. Calculators may be used during examinations. Calculators may not be shared and their cases or covers must be removed. Cell phones may not be used as calculators during examinations. Therefore, they may not be physically present during examinations. During exams, you will be required to leave your books, backpacks, notebooks, etc. at the front of the room. All exams are closed book and closed notes unless otherwise noted. Each exam MUST be signed and this signature will be taken as a statement of honest, independent work When you are finished with your exam, please bring your completed exam to the front, and leave the room quietly without disturbing the other students.

Exams will be graded and returned to you as quickly as possible, usually by the following class. All grading questions, points of clarification, and grading errors must be brought to the instructor's attentions during office hours no later than one week after return of the exam.

The grading scale used to determine letter grades is as follows: **A** 100 – 93, **A-** 92 – 86, **B+** 85 – 82, **B** 81 – 78, **B-** 77 – 74, **C+** 73 – 70, **C** 69 – 65, **C-** 64 – 62, **D** 61 – 50, **F** < 50.

3. *MasteringChemistry Homework (ZOSNERCHEM102FALL2013):* There will be MasteringChemistry homework sets for each chapter we cover, for a total of 60 points. Only 50 points will count towards your final grade. Any points earned over 50 will be counted as extra credit. Late assignments will not be counted for credit.

4. *Discussion:* The discussion section will be devoted to working on discussion hand-outs and answering questions regarding homework problems. *Attendance and participation are mandatory and worth 4 points per class.* There are 14 discussion classes, but you will receive attendance points for 12 classes. If you miss a class, you will not be able to make up any lost points. If you attend ALL 14 classes, you will earn 8 extra credit points. No make-up discussion points will be given under any circumstances.

5. *Sakai Materials*: Handouts given in class will be mirrored on Sakai.

6. *Academic Honesty:* For this course, all exams are closed book and closed note. Academic dishonesty includes using notes or books during exams, looking at another student's test during the exam period, or talking during an exam. The consequence of academic dishonesty ranges from: is failure of the exam (which cannot be dropped), and the incident will be reported to the Chemistry Department Chair and the Office of the Dean to failure of the course. Additional sanctions including expulsion from the university may be imposed. The Undergraduate Handbook contains a complete description of the University policy regarding academic dishonesty.

7. *Office Hours:* My office door will be open per the times listed. Please use this time to if you have extra questions regarding this course. If you are unavailable to meet at the listed times, please feel free to email me with any questions. However, if you email me at night (after 6:00 p.m.), on weekends, or during holiday breaks I will respond to your email within 12 hours.

8. *Students with Disabilities Policy:* Eligibility for services is determined on an individual basis based on documented need. Self-disclosure and the submission of documentation can be initiated anytime during the year. However, reasonable time must be allowed before the student can expect accommodations to be in place. Self-disclosure and documentation are required only if students plan to request accommodations. Students should provide information and documentation at a reasonably early date to allow time for the development and arrangement of appropriate accommodations. In some cases, several weeks' advance arrangement is needed. Accommodations cannot be retroactive. Accommodations begin only after documentation is received and reasonable time for accommodation development has been allowed. http://www.luc.edu/sswd/index.shtml

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General Chemistry 102 Tentative Lecture Schedule (subject to change)

8-27	11	Liquids and Intermolecular Forces
8-29	11/13	Liquids and Intermolecular Forces/Properties of Solutions
9-3	13	Properties of Solutions
9-5	13	Properties of Solutions
9-10	14	Chemical Kinetics
9-12	14	Chemical Kinetics
9-17	14	Chemical Kinetics
9-19	--	**EXAM I (Chapters 11, 13 & 14 or as announced)**
9-24	15	Chemical Equilibrium
9/26	15	Chemical Equilibrium
10-1	15/16	Chemical Equilibrium/Acid-Base Equilibria
10-3	16	Acid-Base Equilibria
10-8	--	*Fall Break (no class)*
10-10	16	Acid-Base Equilibria
10-15	17	Additional Aspects of Aqueous Equilibria
10-17	17	Additional Aspects of Aqueous Equilibria
10-22	17	Additional Aspects of Aqueous Equilibria
10-24	--	**EXAM II (Chapters 15-17 or as announced)**
10-29	19	Chemical Thermodynamics
10-31	19	Chemical Thermodynamics (*Happy Halloween*)
11-5	19/20	Chemical Thermodynamics/Electrochemistry
11-7	20	Electrochemistry
11-12	20	Electrochemistry
11-14	20/21	Electrochemistry/Nuclear Chemistry
11-19	21	NuclearChemistry
11-21	--	**EXAM III (Chapters 19-21 or as announced)**
11-26	23	Transition Metals and Coordination Chemistry
11-28	--	*Thanksgiving Break (no class)*
12-3	23	Transition Metals and Coordination Chemistry
12-5	23	Transition Metals and Coordination Chemistry
12-14	--	**FINAL EXAM CUMULATIVE** **9:00-11:00 a.m.**